

LABDANE DITERPENE GLYCOSIDES WITH 6-DEOXY-L-IDOSE FROM *ASTER SPATHULIFOLIUS* MAXIM.

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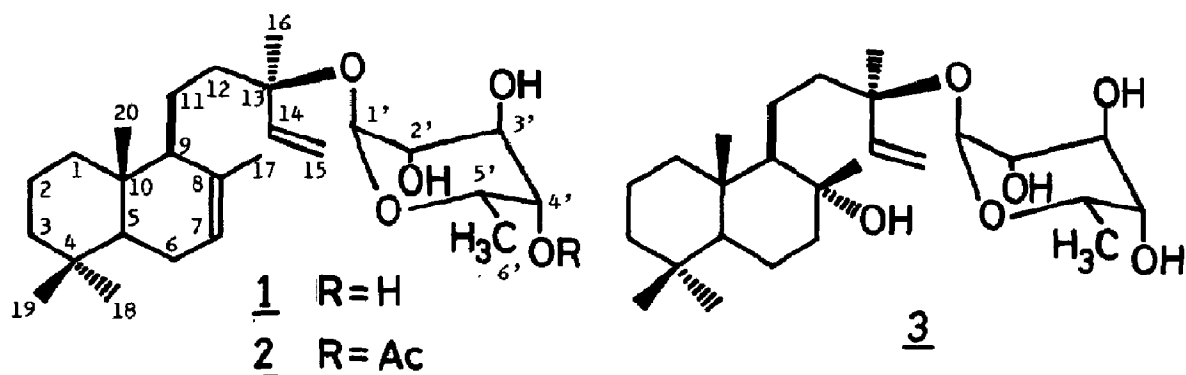
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Summary: Three labdane type diterpene glycosides (1, 2 and 3) with 6-deoxy-L-idose have been isolated from *Aster spathulifolius* Maxim. and their structures have been determined on the basis of chemical and spectral data.

Among eight theoretically possible pairs of optically active 6-deoxyhexoses only 6-deoxyidose has not been known² in nature. We now isolated the sugar as a common component of three labdane diterpene glycosides during our studies on diterpenoids of *Aster spathulifolius* Maxim. The present communication deals with the isolation and structure of these glycosides containing 6-deoxyidose.

Methanolic extract of the fresh aerial parts of the plant was chromatographed on silica gel using chloroform with increasing amounts of methanol, and three closely related glycosides (1, 2 and 3) were isolated along with three diterpenoids, labda-7,14-dien-13 β -ol (7),³ 7 α -hydroxymanool⁴ and sclareol (8).⁵

The glycoside 1,⁶ C₂₆H₄₄O₅ (M⁺ 436), amorphous powder, { α]_D -61.0°(MeOH), was characterized as a 6-deoxyhexoside of a diterpene alcohol based on 26 carbon signals of the ¹³C NMR spectrum (Table 1). Oxidation of 1 with sodium periodate in EtOH furnished a diterpene alcohol, C₂₀H₃₄O (M⁺ 290); { α]_D -2.4°(MeOH), coinciding with 7 isolated from this plant in spectral data (IR, ¹H NMR, ¹³C NMR, MS) and { α]_D-. Acetylation of 1 with Ac₂O/Pyridine gave a tri-O-acetyl derivative (4), C₃₂H₅₀O₈ (M⁺ 582); { α]_D -45.2°(CHCl₃); δ 2.06 (6H, s), 2.10 (3H, s), and also methylation with CH₃I/NaH-DMSO⁷ gave a tri-O-methyl derivative (6), C₂₉H₅₀O₅ (M⁺ 478); { α]_D -38.1°(CHCl₃); δ 3.40, 3.52, 3.54 (each 3H, s). The ¹H NMR spectra of 1, 4 and 6 showed the presence of a secondary methyl and a proton on a methine group forming an O-glycoside linkage at δ 1.20 (3H, d, J=7.0 Hz) and 4.42 (1H, dq, J=7.0, 3.5), 1.18 (3H, d, J=7.0) and 4.42 (1H, dq, J=7.0, 3.5), and 1.16 (3H, d, J=7.0) and 4.15 (1H, m), respectively. From these chemical and spectral data the structure of 1 was confirmed to be a 6-deoxypyranoside of 7. Six ¹³C NMR resonances of sugar moiety of 1 clearly differed from those of reported commonly occurring 6-deoxysugars (rhamnose, fucose, quinovose) and their methyl glycosides.⁸ In addition to the above fact, no formation of acetonide of 1 strongly suggested the presence of 6-deoxyidopyranosyl moiety, the three hydroxyl groups of which were situated in *trans* relationship each other in the molecule of 1. Therefore, a search has been made to isolate the sugar moiety of 1. Methanolysis of 1 using a strongly acidic ion exchange resin⁹ and subsequent acetylation (Ac₂O/Pyridine) afforded two methyl 6-deoxyhexoside triacetates,¹⁰ C₁₃H₂₀O₈ (M⁺-1, 303), oil, { α]_D -148.0°(CHCl₃), and C₁₃H₂₀O₈ (M⁺-1, 303), mp 97.5-98.0°, { α]_D +46.9°(CHCl₃), which were assigned to the α ¹¹ and β -anomeric pyranosides, respectively, based on the ¹H chemical shifts¹² of each secondary methyls (δ 1.20 in



α -anomer and δ 1.30 in β -anomer, each d, $J=7.0$ Hz). The β -anomer obtained was identified to be methyl 2,3,4-tri-O-acetyl- β -L-ido-pyranoside by the mixed mp, $[\alpha]_D$ and spectral (IR, ^1H NMR, MS) comparison with those of synthetic material,¹³ thus confirming the occurrence of 6-deoxy-L-ido-pyranose in the molecule of 1

Table 2. Comparative ^1H NMR data^a of sugar moieties of 1, 2 and 3 in CD_3OD (δ from TMS, 90 MHz).

<u>1</u>	Coupling (Hz)	<u>2</u>	Coupling (Hz)	<u>3</u>	Coupling (Hz)	
H-1'	4.83 (d)	$J_{1,2}=3.5$	4.78 (d)	$J_{1,2}=4.5$	4.81 (d)	$J_{1,2}=4.5$
H-2'	3.35-3.45 (overlaps with H-4', m)	$J_{2,3}=J_{3,4}=5.5$	3.39 (dd)	$J_{2,3}=J_{3,4}=7.0$	3.38 (dd)	$J_{2,3}=J_{3,4}=-6.0$
H-3'	3.69 (t)		3.68 (t)		3.68 (t)	
H-4'	3.35-3.45 (overlaps with H-1', m)	$J_{4,5}=3.5$	4.76 (dd)	$J_{4,5}=4.0$	3.46 (dd)	$J_{4,5}=3.5$
H-5'	4.24 (dq)	$J_{5,6}=7.0$	4.32 (dq)	$J_{5,6}=7.0$	4.24 (dq)	$J_{5,6}=7.0$
H-6'	1.20 (d)		1.13 (d)		1.19 (d)	
CH_3CO			2.08 (s)			

^aThe assignment was made by comparison with the spectrum of 6-deoxy- β -L-ido-pyranoside and its triacetate in ref. 13.

The glycoside 2,⁶ $\text{C}_{28}\text{H}_{46}\text{O}_6$ (M^+ 478), amorphous powder, $[\alpha]_D -46.8^\circ$ (MeOH), which contained an acetoxy group (^1H : δ 2.08, 3H, s, in CD_3OD . ^{13}C : δ 170.4, s, δ 20.7, q, in $\text{C}_5\text{D}_5\text{N}$), was proved to be a monoacetate of 1 since ^{13}C NMR signals of aglycone moieties of 1 and 2 appeared at almost the same positions as listed in Table 1 and on acetylation ($\text{Ac}_2\text{O}/\text{Pyridine}$) 2 gave a triacetate coinciding with 4 in spectral data (IR, ^1H NMR, ^{13}C NMR, MS) and $[\alpha]_D$. Comparison of the ^1H NMR spectra of 1 and 2 (the signals of only sugar parts were listed in Table 2), and decoupling experiment revealed the site of the acetoxy group of 2 to be at C-4'; the signals due to H-2' and H-3', respectively, appeared at δ 3.39 (dd) and δ 3.68 (t) which were essentially the same positions as those of 1, while the signal of H-4' being deshielded on acetylation shifted to a lower field of δ 4.76 which on irradiation caused the triplet (δ 3.68, H-3') into a doublet ($J=7.0$ Hz).

The minor component, glycoside 3,⁶ $\text{C}_{26}\text{H}_{46}\text{O}_6$ (M^+ -18, 436), $[\alpha]_D -33.2^\circ$ (MeOH), on acetylation ($\text{Ac}_2\text{O}/\text{Pyridine}$) provided a tri-O-acetyl derivative (5), $\text{C}_{32}\text{H}_{52}\text{O}_9$ (M^+ 580), $[\alpha]_D -20.2^\circ$ (CHCl_3); δ 2.04 (6H, s), δ 2.06 (3H, s) in which one hydroxyl group remained unaffected (IR, ^1H NMR and (M^+ -18 peak) indicating its tertiary nature. Comparative study of the ^{13}C and ^1H NMR spectra of

Table 1. ^{13}C chemical shifts^a and glycosylation shifts^b of aglycone moieties.

	in $\text{C}_5\text{D}_5\text{N}$			in CDCl_3				
	$\underline{1}^c$	$(\Delta\delta)^b$	$\underline{2}^c$	$(\Delta\delta)^b$	$\underline{3}^c$	$(\Delta\delta)^b$	$\underline{4}^c$	$\underline{5}^c$
C-1	39.3	(0.0)	39.3	(0.0)	39.9	(-0.1)	39.2	39.8
2	19.0	(-0.1)	19.0	(-0.1)	18.7	(-0.1)	20.4	18.9
3	42.5	(0.0)	42.5	(0.0)	42.1	(-0.1)	42.4	42.1
4	33.0	(0.0)	33.0	(0.0)	33.2	(-0.1)	32.9	33.2
5	55.4	(0.0)	55.4	(0.0)	56.3	(-0.2)	55.2	56.2
6	24.0	(0.0)	24.0	(0.0)	20.7	(-0.2)	23.8	20.7
7	122.3	(+0.2)	122.3	(+0.2)	44.3 ^e	(-0.8)	122.4	43.8 ^e
8	135.4	(-0.3)	135.5	(-0.2)	73.8	(+0.3)	135.4	74.3
9	50.3	(-0.1)	50.3	(-0.1)	62.1	(0.0)	50.2	62.0
10	37.2	(-0.1)	37.2	(-0.1)	39.4	(-0.1)	37.1	39.1
11	γ 21.3	(-0.4)	21.4	(-0.3)	19.7	(-0.4)	21.0	20.7
12	β 44.9	(-0.9)	44.9	(-0.9)	44.1 ^e	(-2.3)	44.7	42.8 ^e
13	α 80.4	(+7.7)	79.9	(+7.2)	80.5	(+7.4)	80.4	80.2
14	β 143.0	(-3.8)	143.8	(-3.0)	144.2	(-3.4)	142.1	143.0
15	γ 111.5	(+4.2)	115.0	(+3.8)	114.1	(+3.3)	116.0	114.4
16	β 22.7	(-5.8)	23.2	(-5.3)	23.1	(-5.1)	22.3	23.6
17	22.4	(-0.1)	22.4	(-0.1)	24.3	(-0.3)	22.3	24.3
18	33.3	(0.0)	33.2	(-0.1)	33.4	(-0.2)	33.2	33.2
19	21.9	(0.0)	21.9	(0.0)	21.6	(-0.1)	21.9	21.5
20	13.8	(+0.1)	13.8	(+0.1)	15.0	(-0.6)	13.6	14.9
C-1'	96.6	(-)	96.9	(-)	95.8	(-)	93.2	92.5
2'	71.9 ^d	(-)	73.8 ^d	(-)	72.6 ^d	(-)	69.3 ^d	69.7 ^d
3'	72.5 ^d	(-)	71.8 ^d	(-)	72.8 ^d	(-)	70.1 ^d	70.1 ^d
4'	71.6 ^d	(-)	75.9	(-)	72.5 ^d	(-)	68.8 ^d	68.9 ^d
5'	65.6	(-)	65.8	(-)	67.3	(-)	63.3	64.2
6'	15.7	(-)	14.3	(-)	15.6	(-)	15.1	15.5
			Ac {	170.4	(-)			
				20.7	(-)			
						3 x Ac {	170.0	169.9
							169.5	169.6
							169.3	169.3
							20.8	20.7

^aThe spectra were taken on a Hitachi R-42FT spectrometer at 22.6 MHz and reported in ppm relative to TMS; assignment were aided by off-resonance decoupling of each compound.

^bChange in carbon chemical shift on glycosylation¹⁴: $\Delta\delta = \delta(\text{glycoside}) - \delta(\text{aglycone})$, in ppm.

^cThe assignment of aglycone moieties referred to labdane diterpenoids.¹⁷ ^{d,e}Values within a column may be interchanged.

3 with those of 1 and 8 co-occurred in the same plant enabled us to determine the structure of 3. Namely, ^{13}C signals due to both aglycone and sugar moieties of 3 resonated at virtually the same positions as those of 1, except for the signals of C-7 and C-8, as well as those of 4 and 5 (Table 1), and a good agreement was also obtained in the ^1H chemical shifts and J values of sugar parts of 1 and 3 (Table 2). Taken together, these data supported the structure of 3 which was a glycoside of sclareol (8) containing 6-deoxy-L-idose as a sugar component. The attachment of the sugar moiety to the *tert*-OH at C-13 of 3 could be established by difference of the ^{13}C chemical shifts with those of corresponding aglycone (8); characteristic signal shifts¹⁴ were observed in the α , β and γ -positions of the OH group in which the glycosylation occurred as well as those of 1 and 2, as shown in parentheses of Table 1. Lastly, the anomeric proton doublets of 1, 2 and 3 showed respectively the coupling constants of 3.5, 4.5 and 4.5 Hz, these values of which were closely similar to those of methyl α -idopyranosiduronic acid (4.0 Hz)¹⁵ and methyl α -D-idopyranoside (4.0 Hz).¹⁶

Based on the results mentioned above, the structures of three glycosides isolated from *A. spathulifolius* have been represented by formulae 1, 2 and 3, whose anomeric carbons were determined to be all in α configurations.

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REFERENCES AND NOTES

1. The present address: Department of Chemistry, Faculty of Science, Okayama University of Science, Okayama 700, Japan.
2. T. K. Devon and A. I. Scott, "Handbook of Naturally Occurring Compounds," Vol. I, Academic Press, New York and London (1975); "Rodd's Chemistry of Carbon Compounds," 2nd ed, ed by S. Coffey, Elsevier Publishing Co., Amsterdam, London and New York (1967), Vol. 1, Part F, P500.
3. R. M. Carman, W. J. Craig and I. M. Shaw, *Aust. J. Chem.*, **26**, 215 (1973). We are grateful to Prof. R. M. Carman, University of Queensland, for the ^1H NMR and IR spectra of the specimen.
4. R. C. Cambie, P. K. Grant, C. Huntrakul and R. J. Weston, *Aust. J. Chem.*, **22**, 1691 (1969). We thank Prof. R. C. Cambie, University of Auckland, for the identification of our sample.
5. (8): mp 104.5-105.0°, $[\alpha]_D^{25} = -5.1^\circ$ (CHCl_3). Treatment of 8 with $\text{Ac}_2\text{O}/4$ -dimethylaminopyridine afforded di-O-acetyl-(-)-sclareol, mp 79.5-80.0° (petroleum ether), $[\alpha]_D^{25} = -30.0^\circ$ (CHCl_3), coinciding with the reported data: C. Ohloff, *Helv. Chim. Acta*, **41**, 845 (1958).
6. Satisfactory microanalyses or high mass measurements were obtained.
7. S. Makomori, *J. Biochem. (Tokyo)*, **55**, 205 (1964).
8. P. A. J. Gorin and M. Mazurek, *Can. J. Chem.*, **53**, 1212 (1975).
9. T. Takita, K. Maeda, H. Umezawa, S. Omoto and S. Umezawa, *J. Antibiot.*, **22**, 237 (1969).
10. Separation was performed by preparative TLC on silica gel using petroleum ether-ethyl ether (v/v, 1:1); Rf: 0.36 (α -anomer) and 0.21 (β -anomer).
11. ^{13}C NMR data (CDCl_3); CH_2 : 15.6; CH : 62.8, 68.0, 68.6, 69.9; anomeric CH : 99.2; CH_2CO : 20.7 ($\times 3$); CH_3CO : 170.0 ($\times 2$), 169.4; CH_3O : 55.5 ppm. The spectrum exhibited a good correspondence with those of sugar part of 4 listed in Table 1, except for the signals due to C-1' and CH_3O .
12. H. B. Sinclair and R. T. Sleeter, *Tetrahedron Letters*, 833 (1970).
13. D. Ikeda, T. Tsuchiya and S. Umezawa, *Bull. Chem. Soc. Jpn.*, **44**, 2529 (1971). We thank Prof. S. Umezawa, Institute of Bio-organic Chemistry, Kawasaki, Japan, for supply of the synthetic material.
14. R. Kasai, M. Suzuo, J. Asakawa and O. Tanaka, *Tetrahedron Letters*, 175 (1977); K. Tori, S. Seo, Y. Yoshimura, H. Arita and Y. Tomita, *ibid.*, 179 (1977).
15. A. S. Perlin, B. Casu, G. R. Sanderson and J. Ise, *Carbohyd. Res.*, **21**, 123 (1972).
16. S. J. Angyal and V. A. Pickles, *Aust. J. Chem.*, **25**, 1695 (1972).
17. B. L. Buckwalter, I. R. Burfitt, A. A. Nagel, E. Wenkert and F. Naf, *Helv. Chim. Acta*, **58**, 1567 (1975); S. O. Almqvist, C. R. Enzell and F. Wehrli, *Acta Chem. Scand.*, **B 29**, 695 (1975).

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